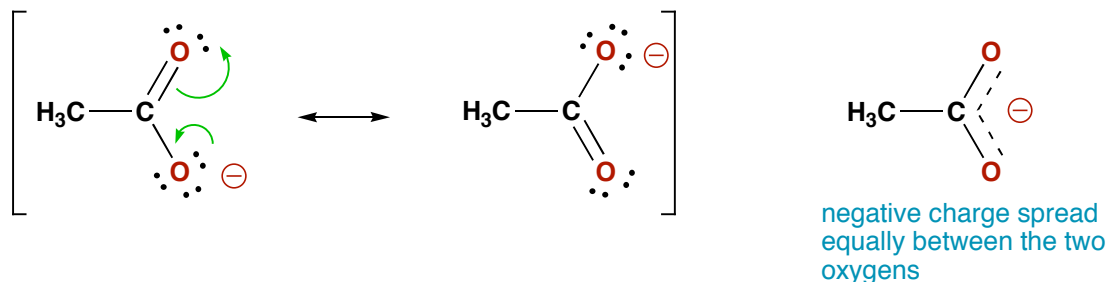


- Individual resonance forms **DO NOT EXIST** - they represent the extremes. The actual chemical species is a hybrid of all the resonance forms.
- Resonance forms differ only in the distribution of pi or non-bonding electrons. Atoms do not move.
- Individual resonance forms must obey normal valency rules. You cannot break the octet rule.
- The actual hybrid is more stable than the individual resonance forms and the individual structures do not have to be equal in energy.

Acetate Ion



Benzene

